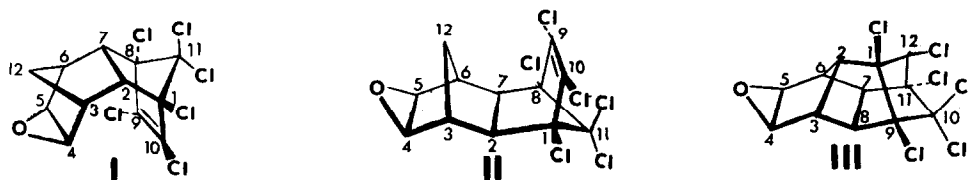


**Eu(DPM)₃ TRANSANNULARLY INDUCED PARAMAGNETIC CHEMICAL SHIFTS IN
THE PMR SPECTRA OF ENDRIN, DIELDRIN, AND PHOTODIELDRIN**

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(Received in USA 24 November 1970; received in UK for publication 30 November 1970)

The new PMR shift reagent tris(dipivalomathanato)europium [Eu(DPM)₃; 1-3] was found useful for the chemical shift and structural assignments of the epoxy derivatives of the chlorinated polycyclodiene pesticides and their degradation products. Eu(DPM)₃ associates with epoxide oxygens and induces large paramagnetic shifts in the signals of protons in adjacent and distant rings. Studies of the effects of varied concentrations of Eu(DPM)₃ on the PMR spectra of the insecticides endrin (I), dieldrin (II), and the important photolysis product of II, photodieldrin (III, 4,5) showed the relationship between the distance of the protons from the europium and the magnitude of the chemical shift.



The PMR spectra of endrin and dieldrin are similar in most respects but one definitive feature of the endrin spectrum (6) is the splitting of the H_{2,7} signal into a pair of doublets due to the coupling of H₂ with H₃ and H₆ and the coupling of H₇ with H₆ and H₃. This characteristic of the endo-endo ring fusion is obscured by the overlap of the H_{2,7} and the H_{4,5} signals of endrin in CCl₄

(Figure 1A). Addition of $\text{Eu}(\text{DPM})_3$ to the endrin solution shifts both the $\text{H}_{2,7}$ and the $\text{H}_{4,5}$ signals downfield (Figure 1B) but the latter shift is approximately four times greater in magnitude so these two signals become well separated and the splitting of the $\text{H}_{2,7}$ signal can be clearly observed at a 250 Hz sweep width. Dieldrin, which possesses an endo-exo ring fusion, does not contain dihedral angles conducive to the above couplings and its $\text{H}_{2,7}$ protons resonate as a sharp singlet (6).

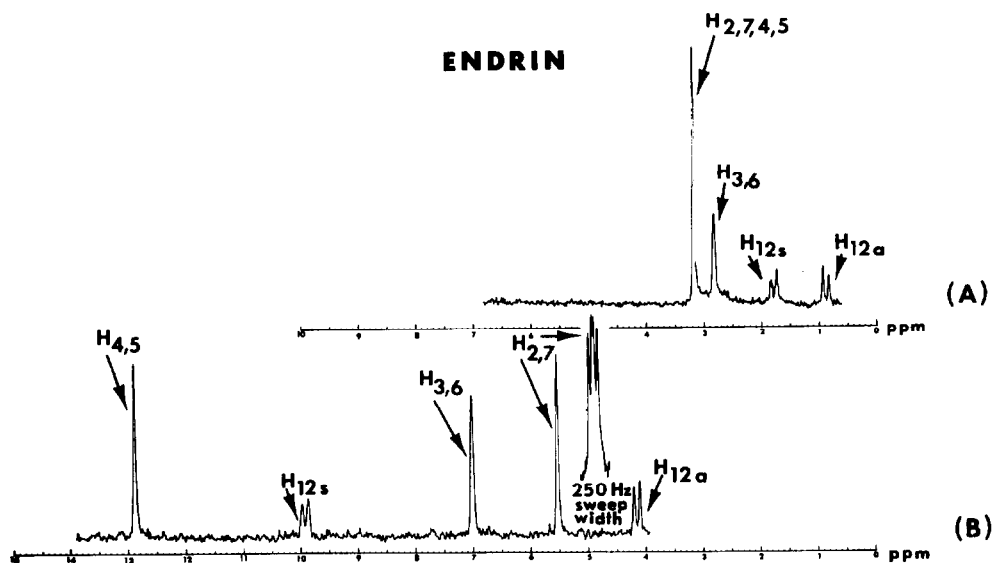
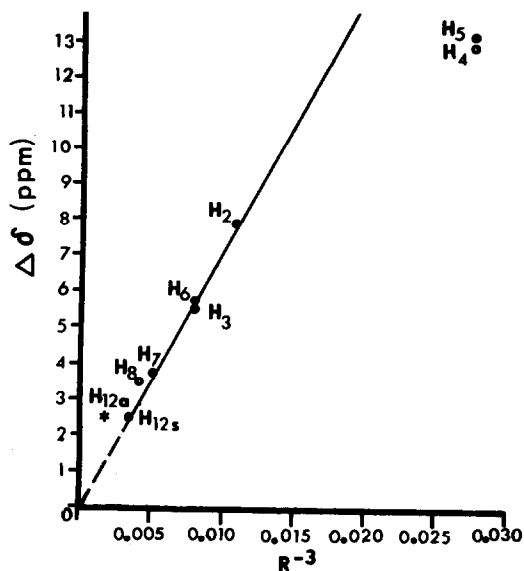


Figure 1 (A) Endrin (0.14 M) without $\text{Eu}(\text{DPM})_3$ and (B) with 0.07 M $\text{Eu}(\text{DPM})_3$ in CCl_4 at 100 MHz.

Graphs of $\Delta\delta$ versus R^{-3} (where R is the distance of each proton from the center of the europium atom (3) as measured from molecular models) give straight lines for all three epoxides. Figure 2 shows the plot obtained for photodieldrin using this method. All except the two protons closest to the europium (H_4 and H_5) fit the curve as might be expected (1).

Figure 2 Change in chemical shift versus the reciprocal of the distance cubed of each proton of a 0.20 M solution of photodieldrin and 0.18 M solution of $\text{Eu}(\text{DPM})_3$ in carbon tetrachloride.



The stereochemistry of H_{12} , which migrates during the photorearrangement reaction, has never been determined to our knowledge. It is seen from Figure 2 that $\Delta\delta$ of this proton plotted for the configuration syn to the epoxide (H_{12s}) conforms well with the curve whereas the anti configuration (H_{12a} ; designated by the asterisk) does not. This indicates that the H_{12} is syn to the epoxide.

Close examination of the H_{12} signal revealed that it is a poorly resolved doublet with a small coupling constant ($J = 0.7$ Hz). Irradiation of either H_2 or H_7 sharpens the doublet perceptibly and triple resonance at these frequencies doubles this effect. Thus, both H_2 and H_7 have a small long-range coupling (<0.5 Hz) with H_{12} . Normally, such coupling is confined to protons separated by four bonds and in a "W" conformation with respect to one another, which would indicate that H_{12} is anti to the epoxide. However, a recent investigation by Marchand *et al.* (7) noted a small, but finite, non-"W" long-range coupling with the proton geminal to the position where this coupling usually occurs. Similarly, we noted this anomaly in the syn-methylene proton signal of those pesticides containing the epoxide or double bond linkage at $\text{C}_4\text{-C}_5$ (6). Irradiation of H_6 collapses H_{12} to a singlet making it evident that these two

protons must be spatially close to one another. Examination of the model of photodieldrin reveals that H₆ and H₁₂ are the "bowsprit" and "flagpole" hydrogens on a six-membered ring rigidly held in a skew-boat conformation (Figure 3). The distance between these two protons is only about 1.6 Å whereas the sum of the van der Waals radii of two hydrogens is 2.4 Å (8) so that a transannular through-space coupling of these protons is the most likely cause of the H₁₂ doublet, confirming the H_{12s} assignment made on the basis of the Δδ produced by Eu(DPM)₃.

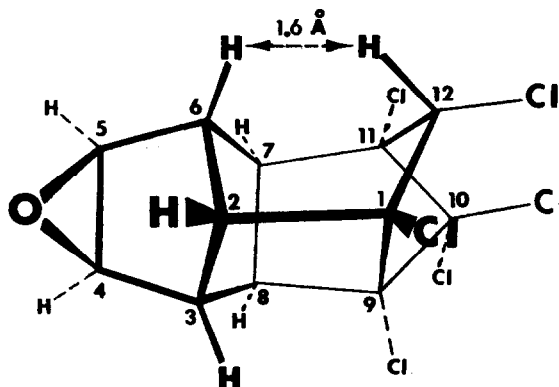


Figure 3 Stereochemical relationship between H₆ and H₁₂ of photodieldrin

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